STEREOSELECTIVE PHOTOSOLVOLYSES OF <u>ipso</u>-acetoxynitro-adducts OF <u>p</u>-XYLENE

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<u>Summary</u>: Diastereoisomers of 1,4-dimethyl-4-nitro-2,5-cyclohexadienyl acetate exchanged acetoxyls for alkoxyls on irradiation in alcohols to give the diastereoisomers of alkyl 1,4-dimethyl-4-nitro-2,5-cyclohexadienyl ethers in isomer ratios of 70:30, whereas the irradiation in pentane yielded p-xylene.

Nitrations at a substituted position (<u>ipso-position</u>) of aromatic hydrocarbons have been studied extensively.¹ Fischer and Ramsay have found acidcatalyzed solvolytic replacements of acetoxyls with methoxyls in the <u>ipso-</u> acetoxynitro adducts of <u>p-xylene</u> (**1a** and **1b**) to give ipso-methoxynitro adducts



(2a and 2b).^{2,3} In this paper we would like to report similar solvolyses induced by photolyses, which showed striking stereoselectivities in marked contrast to the nonstereoselective acid-catalyzed solvolyses.²

The acetoxydienes la and lb were prepared according to the literature.² A <u>cis</u>-configuration in respect to the 1- and 4-methyl groups has tentatively been assigned to la and a <u>trans</u>-one to lb.² Irradiation of la (100 mg) in 10 ml of methanol under nitrogen at 20 °C with an external 100 W high pressure mercury

lamp equipped with a Pyrex filter yielded an oil including 70% of 2a and 30% of 2b as revealed by NMR. The combined yield of 2a and 2b based on the converted la was almost quantitative through a conversion of about 95%. Similar irradiation of 1b gave 30% of 2a and 70% of 2b. The IR and NMR spectra of the products coincided with those of the authentic 2a or 2b which were prepared by the solvolysis of the acetoxydiene catalyzed by sulfuric acid.² The configurations of the methoxydienes 2a and 2b have recently been determined; that is, 2a has a <u>trans</u>-configuration and 2b has a <u>cis</u>-one.⁴ The <u>cis/trans</u> ratios of the photoproducts were kept constant throughout the reaction proceedings. In fact 2a and 2b underwent no mutual exchange when irradiated in methanol. The acetoxydienes recovered from the reaction mixtures contained no mutually exchanged isomers, either. The photolyses of the methoxydienes in acetic acid caused no conversion to the acetoxydienes.

Benzyl acetates with methoxyl groups at the meta-position are known to exchange acetoxyls for methoxyls on irradiation in methanol.^{5,6} No stereoselectivity has been found in this system; the photolysis of a chiral benzyl acetate, for example, resulted in racemization.⁶

The photosolvolyses of the benzyl acetates were considered to involve the intermediates of benzyl cation type.⁶ Similarly, the cyclohexadienyl cation (3)



was proposed as an intermediate for the acid-catalyzed solvolysis of the acetoxydiene.² This planar cation cannot account for the stereoselective reaction. The photosolvolysis of the acetoxydiene suggests a synchronous mechanism.

Irradiations of la in ethanol, propanol and allyl alcohol yielded ethyl, propyl and allyl ethers of 1,4-dimethyl-4-nitro-2,5-cyclohexadienol (4, 5 and 6), respectively, in isomer ratios of 70:30; lb underwent similar photosolvolyses affording the same alkoxydienes with reversed isomer ratios. The major photo-

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products from la (2a, 4a, 5a and 6a) were eluted from aluminum columns faster than the corresponding minor products (2b, 4b, 5b and 6b), suggesting a common configuration in a-isomers and the compliment in b-isomers.

The alkoxydienes 4, 5 and 6 were also prepared from 1 by the acid-catalyzed solvolyses by sulfuric acid in the corresponding alcohols. Like the reaction in methanol,² no stereoselectivity was observed in these acid-catalyzed reactions.

Irradiations of the acetoxydienes in pentane or ethyl ether yielded p-xylene. It has been known that the carboxylic esters of aliphatic alcohols are



reduced to alkanes on irradiation in a hexamethylphosphoramide-water system.⁷ Similarly photolysis of 3,5-dimethoxybenzyl acetate in hexane has given rise to a homolytic cleavage of the benzyl-acetate bond.⁸ The photoreactions of the acetoxydienes in the aprotic solvents might be initiated by a similar homolytic cleavage of C^1 -O bond to yield the cyclohexadienyl radical (7) which would subse quently lose a nitro radical to resume the benzene system. Otherwise, as encoun-

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tered in aliphatic nitro compounds,⁹ the photolysis might induce a cleavage of the C-N bond to give another cyclohexadienyl radical (8) as a precursor of p-xylene.

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